

## **S<sub>N</sub>Ar reactivity of thiophene derivatives investigated using conceptual DFT**

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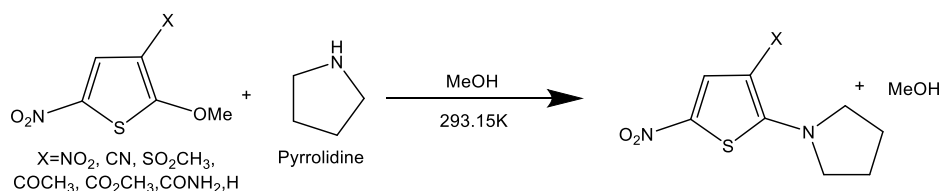
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Thiophenes are heterocycles of industrial interest which have attracted considerable attention due to their versatility in different applications. For instance, substituted thiophenes are used to synthesize materials with photophysical properties and electrochemical systems and they are increasingly being employed in therapeutical applications.[1], [2] The formation of amino substituted thiophenes through aromatic nucleophilic substitution (S<sub>N</sub>Ar) have been extensively studied experimentally.[3], [4] To gain more insights on the substituent effects, the electrophilicities of numerous thiophenes derivatives have been provided using Mayr's electrophilicity scale.[5] However, further analysis of such properties and computational efforts to achieve a description of the reaction mechanism are rare.[6]

In this work, we performed quantum chemistry calculations to investigate the mechanism of the S<sub>N</sub>Ar reaction of pyrrolidine with seven substituted thiophenes.



Scheme of the studied S<sub>N</sub>Ar reaction.

We then used conceptual DFT and Topological approaches [7] to rationalize the evolution of the computed activation energies by means of quantum descriptors, such as Parr's electrophilicity and the LUMO energy. We showed that, though the relative reactivity of different thiophene sites is hard to model, the reactivity of the thiophene derivatives nicely correlates with global descriptors.

### References

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