Investigating organic molecules at the water-platinum interface: From single-crystal surfaces to nanoparticles via density functional tight binding

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The platinum/water interface is a prototype to investigate the impact of transitioning from gas-phase petrochemistry to aqueous-phase biomass valorization. The aqueous environment is notoriously difficult to capture for computational methods:[1] implicit solvent models fail to capture near chemisorption on noble-metal surfaces, and DFT-based molecular dynamics are too costly in view of the slow diffusion constants at the solid/liquid interface. Even though QM/MM for the evaluation of the solvation free energy using dedicated force fields such as GAL17 leads to gualitative agreement with experiment.[2] To go beyond additive force fields, we introduce a density functional tight binding (DFTB) parametrization for the Pt/C/H/O system. We quantify the accuracy describing the adsorption and reaction of polyols on Pt(111) as well as the dynamic (morphological) changes of nanoparticles (Pt₃₈, Pt₅₅ and Pt₁₄₇) to be around 10 kcal/mol. Our parametrization efforts show that DFTB requires a corrective three-body potential, introduced via ChIMES[3] to capture water adsorption on Pt(111) and for the structure and reactivity related to nanoparticles. We showcase the difference between DFTB and DFTB/ChIMES on the adsorption energy of phenol on Pt(111) in the gas-phase and in water, see Figure 1.

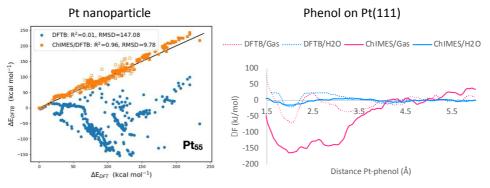


Figure 1 Comparison of DFTB and DFTB/ChIMES.

References

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